

Jawapan Module Teaching Learning Chemistry Form 4 Nilam Publication Pdf

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(iii) Nitrogen molecule / Molekul nitrogen:

- (a) Nitrogen atom with an electron arrangement 2.5 needs 3 electrons to achieve stable octet electron arrangement. / Atom nitrogen dengan susunan elektron 2.5 memerlukan 3 elektron untuk mencapai susunan elektron oktet yang stabil.
- (b) Two nitrogen atoms share 3 pairs of electrons to achieve a stable octet electron arrangement, form a nitrogen molecule. Each nitrogen atom achieves stable octet electron arrangement.
 Dua atom nitrogen berkongsi 3 pasang elektron untuk mencapai susunan elektron oktet yang stabil, membentuk satu molekul nitrogen. Setiap atom nitrogen mencapai susunan elektron oktet yang stabil.

Draw the electron arrangement of the molecule formed. / Lukisk an susunan elektron bagi molekul yang terbentuk.





Steps to Explain the Formation of Covalent Bond for Covalent Compound

Langkah untuk Menerangkan Pembentukan Ikatan Kovalen dalam Sebatian Kovalen

Predict formula / Meramal formula

Determine the formula of the covalent compound formed.

→ Draw the electron arrangement for the compound formed based on the formula. I Lukis susunan elektron untuk sebatian yang terbentuk berdasarkan formula.

Tentukan formula sebatian kovalen yang terbentuk. | terbentuk b

- Write the electron arrangement of each atom.
 Tulis susunan elektron untuk setiap atom.
- By referring to the number of valence electrons for each atom, determine the number of electrons needed by each atom to achieve stable duplet/octet electron arrangement.

Dengan merujuk kepada bilangan elektron valens setiap atom, tentukan bilangan elektron yang diperlukan oleh setiap atom untuk mencapai susunan elektron oktet/duplet yang stabil.

- Cross the number of electrons needed to achieve stable duplet/octet electron arrangement. / Silangkan bilangan elektron yang diperlukan untuk mencapai susunan elektron duplet/oktet yang stabil.
- Electrons are shared in pair.

 Electrons are shared in pair.
- Elektron dikong si dalam pasangan. The number of pairs of electrons si
- The number of pairs of electrons shared is based on the number of electrons needed by each atom to achieve stable duplet/octet electron arrangement.

Bilangan pasangan elektron yang dikongsi adalah berdasarkan bilangan elektron yang diperlukan oleh setiap atom untuk mencapai susunan duplet/oktet yang stabil.

(c) State the electron arrangement achieved by each atom in the molecule (stable duplet/octet electron arrangement).

> Nyatakan susunan elektron yang dicapai oleh setiap atom dalam molekul (susunan elektron duplet/ oktet yang stabil).

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(b) Based on the drawing of the electron arrangement, state:

Berdasarkan lukisan susunan elektron, nyatakan:

- * the number of pairs of electrons shared between atoms. / bilangan pasangan elektron yang perlu dikongsi antara atom.
- * the number of electrons of each atom contribute for sharing in the molecule and type of covalent bond formed (single/double). bilangan elektron yang disumbangkan oleh setiap atom untuk dikongsi oleh setiap atom dalam mole ku! dan je nis ikatan kovalen yang terbentuk (tunggal/ganda du a).

a) State the electron arrangement of each atom and the number of electrons needed to achieve stable du plet/octet electron arrangement.

Nyatakan susunan elektron setiap atom dan bilangan elektron yang diperlukan oleh setiap atom untuk mencapai susunan elektron duplet/oktet yang stabil.

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